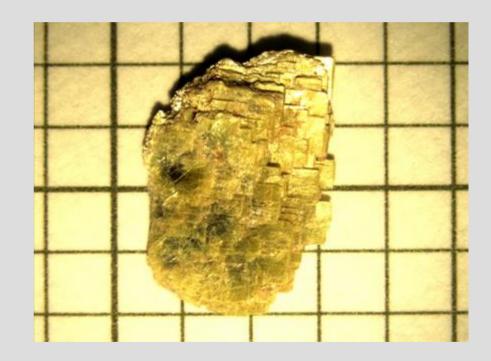
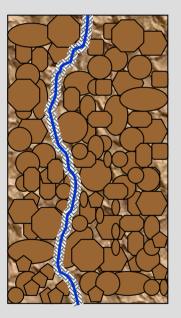


#### **Uranium-Phosphate (Autunite) Minerals**

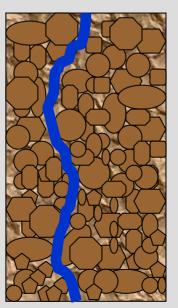
- Very low solubility.
- Formation does NOT depend on changing the redox conditions of the aquifer.
- Not subject to reversible processes such as reoxidation or desorption.



## Challenges to Phosphate Amendments: **Rapid Precipitation Kinetics**



- Injection of monophosphate molecules results in rapid flocculation and precipitation of phosphate phases
- Sharp decrease in hydraulic conductivity.



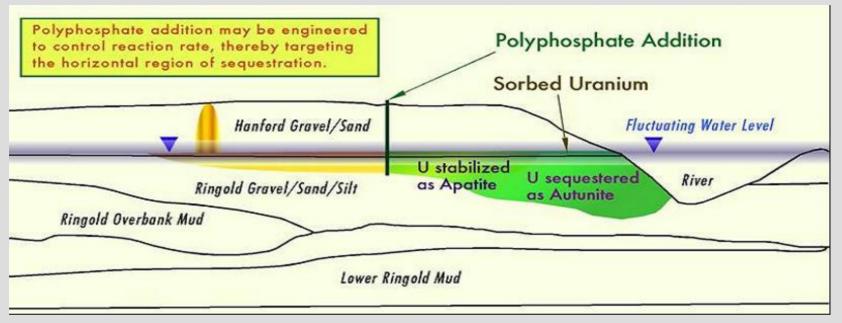
- Polyphosphate precludes rapid precipitation
- No measurable decrease in hydraulic conductivity

## Solution to Deployment Challenges: **Use of Long-Chain Polyphosphates**

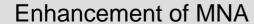
- Slow reaction with water to yield orthophosphate
- Rate of hydrolysis is related to chain length
  - Time release Controllable kinetics based on to polymer length
- Rate of phosphate mineral formation is directly related to the rate of polyphosphate hydrolysis.
  - Direct treatment of uranium
  - Provides immediate and long-term control of aqueous uranium

Polyphosphate amendment can be tailored to delay formation of autunite and apatite.

#### Deployment of Phosphate Amendment for In-Situ Immobilization of Uranium

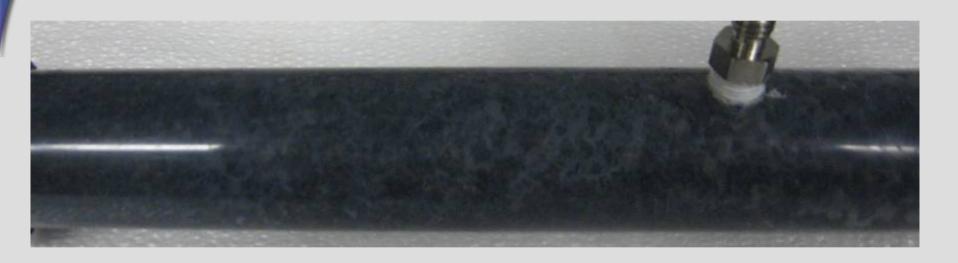


- Injection of soluble polyphosphate
- Lateral plume treatment
- Uranyl phosphate mineral (autunite) formation
  - Immediate sequestration
- Apatite formation
  - Sorbent for uranium
  - Conversion to autunite





#### **Uranium Column Testing**

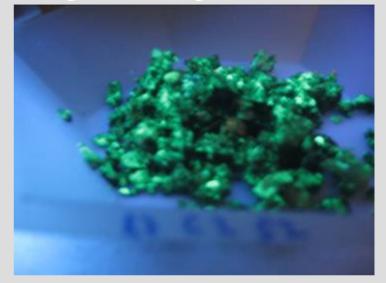


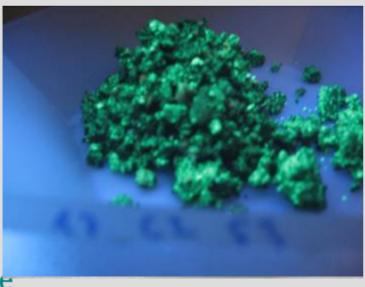
Total 
$$[P]_{aq} = 5.26 \times 10^{-2} \text{ M}$$
  
Pyro  $[P]_{aq} = 6.58 \times 10^{-3} \text{ M}$   
 $[Ca]_{aq} = 9.98 \times 10^{-2} \text{ M}$  pH = 7

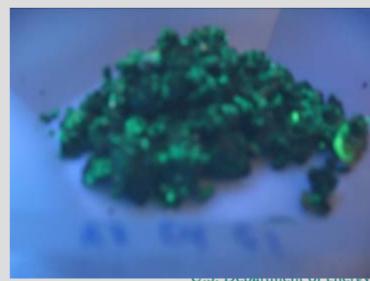
Tripoly 
$$[P]_{aq} = 8.77 \times 10^{-3} \text{ M}$$
  
Ortho  $[P]_{aq} = 1.32 \times 10^{-2} \text{ M}$   
RT = 56 min PV = 52 mL PV = 1 Ca/1P

## **Post-Test Preliminary Analysis**

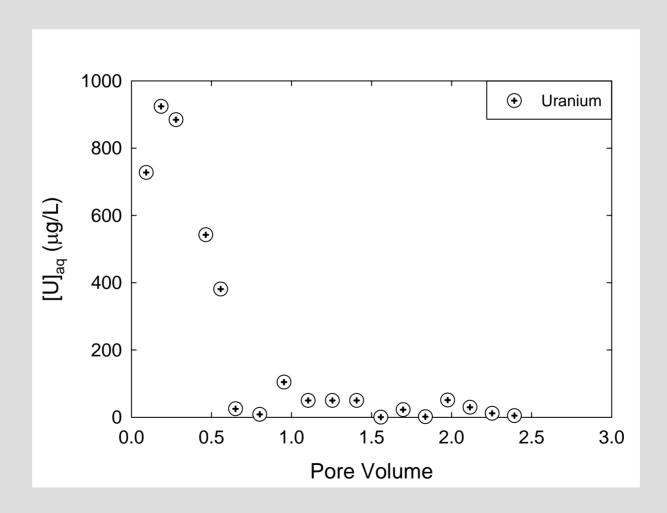








## **Aqueous Uranium During Treatment**



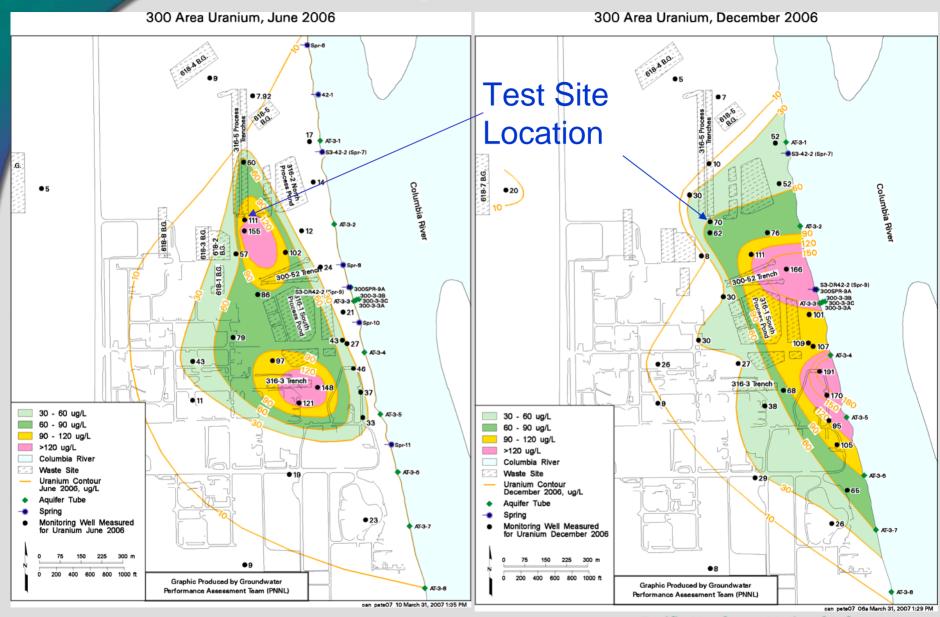


#### **Treatability Testing Activities**

- Bench-scale studies
  - Amendment formulations finalized
  - Phased treatment approach selected
- Site specific characterization
  - Installation of well network
  - Hydrogeologic characterization
  - Hydraulic/tracer injection testing
- Polyphosphate injection design
  - Development of local-scale flow and transport model
  - Determination of injection volumes, rates, and chemical mass requirements
- Polyphosphate injection test
  - Injection conducted in June 07
  - Preliminary performance assessment monitoring

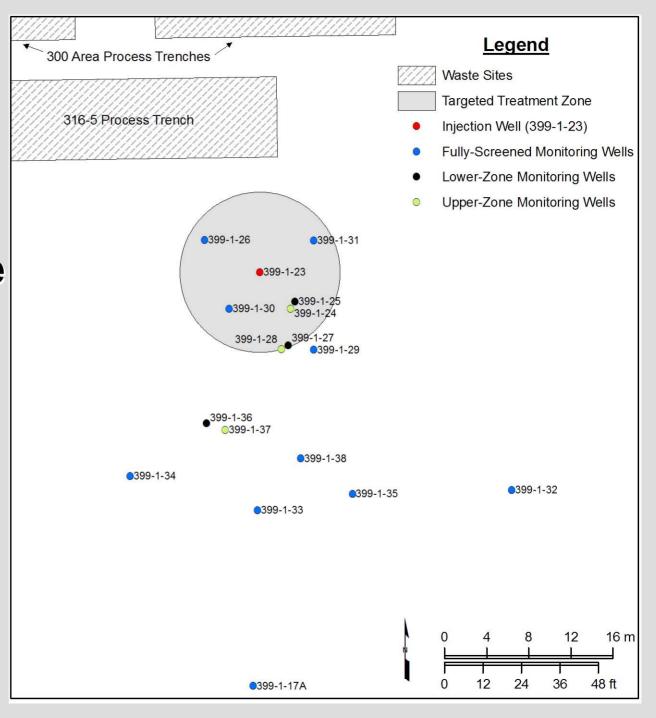


#### **Treatability Test Site Location**

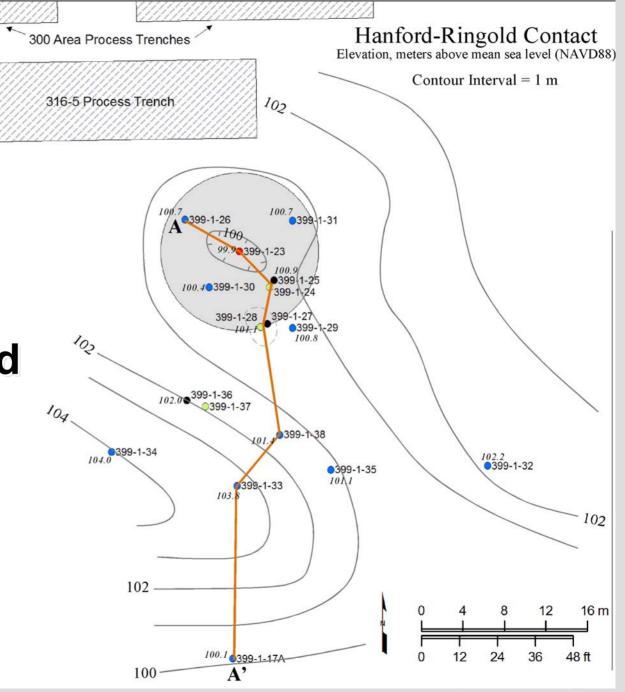




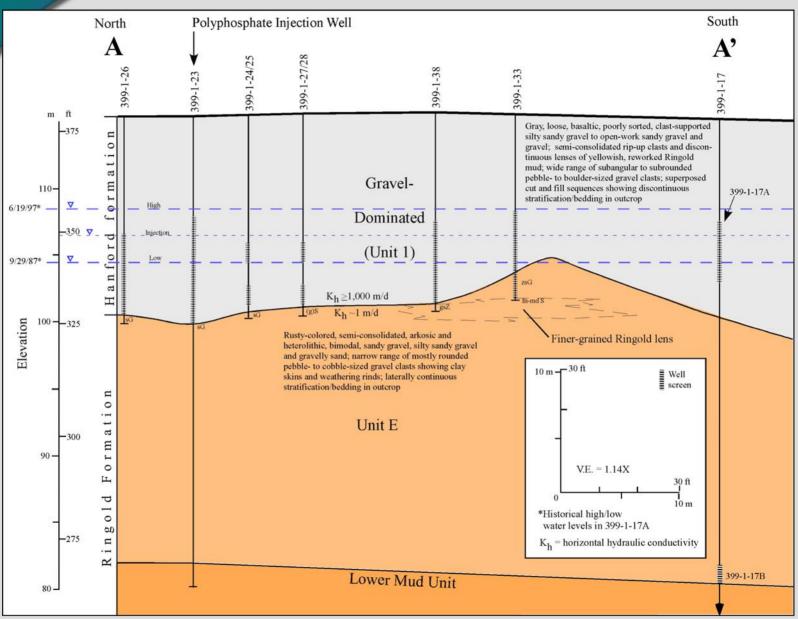
## Polyphosphate Treatability Test site Well Layout



Contour Map Showing Hanford-Ringold Contact



#### **Geologic Cross Section**

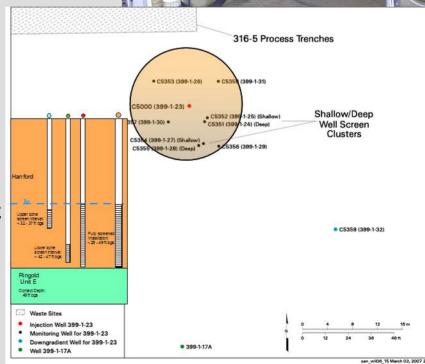




300 Area Polyphosphate Treatability Test Tracer Injection Test

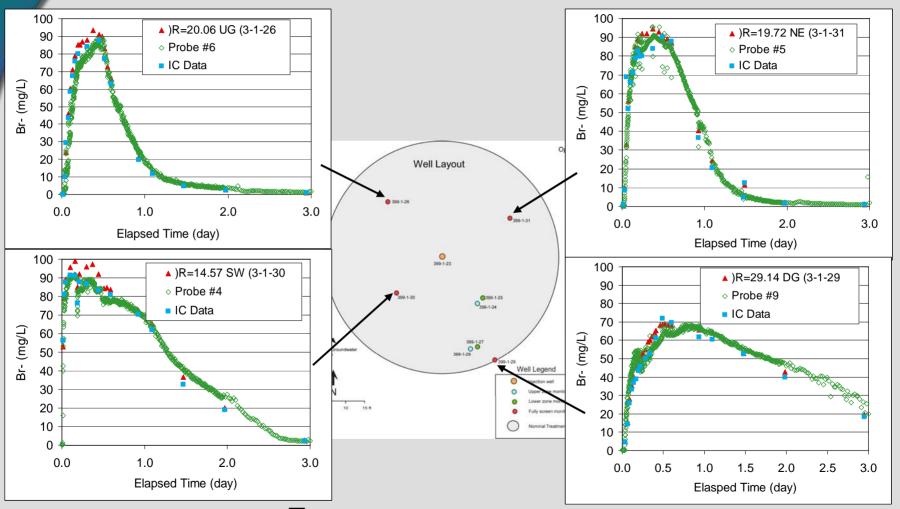
- NaBr tracer test on Dec. 13, 2006
  - Aquifer thickness ~ 15 ft
  - Injection Volume: 143,000 gallons
  - 200 gpm for 11.9 hrs
- ► Inline tracer mixing with water from Well 399-1-7 (620 ft DG)
- ▶ Br⁻ conc. measured in injection stream and surrounding monitoring wells
  - Samples analyzed on site with ISE
  - Archive samples → verification by IC
  - Downhole ISE probes installed in all monitoring wells







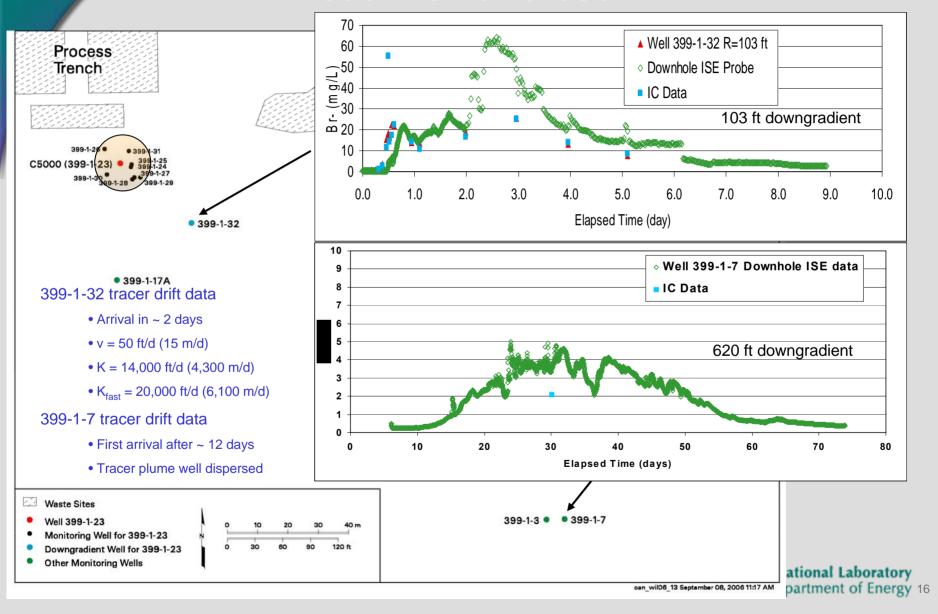
# Tracer Test Results within Targeted Treatment Volume



 $-\overline{n}_{eff}$  (based on tracer arrival)= 0.18

 Consistent with LFI porosity estimates based on physical property analysis

## Tracer Results for Downgradient Wells 399 1-32 and 399-1-7



#### **Treatment Volume Estimation**

- Idealized PV<sub>25 ft</sub> ~ 42,000 gal
- Tracer arrival data normalized to 25 ft radius based on volumetric ratio
- ► Injection volume requirements:

Well Name	Distance to	50% tracer	80% tracer	90% tracer	100% tracer
	399-1-23 (ft)	Arrival (gal)	Arrival (gal)	Arrival (gal)	Arrival (gal)
399-1-23	0.0				_
399-1-24	14.5	77,425	125,072	148,895	339,481
399-1-25	14.1	25,093	50,185	62,731	138,009
399-1-26	20.1	34,175	62,136	86,990	201,940
399-1-27	24.1				
399-1-28	24.3	46,659	95,438	125,130	151,216
399-1-29	29.1	45,640	104,973		
399-1-30	14.6	11,785	17,677	23,569	58,923
399-1-31	19.7	28,941	61,099	77,177	112,550
Average		38,531	73,797	87,415	167,020
Avg. @ high WT		48,292	92,492	109,561	209,332

► Amendment volumes adjusted to account for adsorption: R<sub>f</sub> [PO<sub>4</sub>] ~ 2.4, R<sub>f</sub> [Ca] ~ 4.8

Battelle

#### **Polyphosphate Injection Test**

- Polyphosphate injection on June 11-15, 2007
  - Aquifer thickness ~ 19 ft
- ▶ 3 phase approach: PolyPO<sub>4</sub> / CaCl / PolyPO<sub>4</sub>
  - 200 gpm injection Rate
  - Amendment injection volumes (Kgal): 250 / 500 / 250

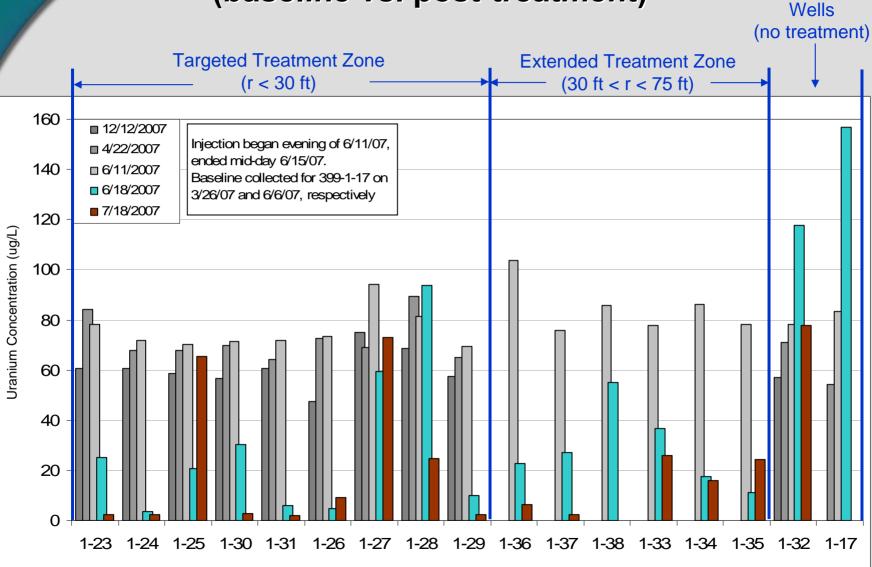
#### ► Amendment Formulation:

Injection	Amendment	Formula	Conc.,	Conc., M
			(g/L)	
1	Sodium Orthophosphate	NaH <sub>2</sub> PO <sub>4</sub>	0.5925	4.94 x 10 <sup>-3</sup>
	Sodium Pyrophosphate	$Na_4P_2O_7$	0.657	$2.47 \times 10^{-3}$
	Sodium Tripolyphosphate	$Na_5P_3O_{10}$	1.209	3.29x 10 <sup>-3</sup>
	Sodium Bromide	NaBr	0.103	$1.00 \times 10^{-3}$
2	Calcium Chloride	CaCl <sub>2</sub>	3.405	3.07 x 10 <sup>-2</sup>
3	Sodium Orthophosphate	NaH <sub>2</sub> PO <sub>4</sub>	0.5925	4.94 x 10 <sup>-3</sup>
	Sodium Pyrophosphate	$Na_4P_2O_7$	0.657	2.47x 10 <sup>-3</sup>
	Sodium Tripolyphosphate	$Na_5P_3O_{10}$	1.209	3.29x 10 <sup>-3</sup>
	Sodium Bromide	NaBr	0.103	$1.00 \times 10^{-3}$

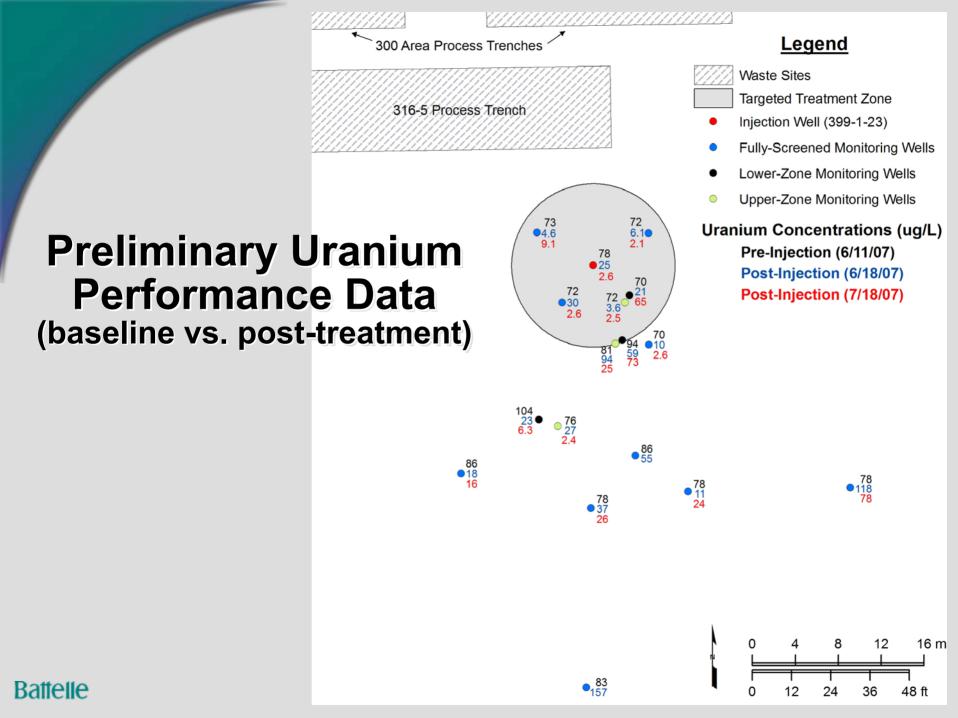


#### Preliminary Uranium Performance Data **Downgradient**

(baseline vs. post-treatment)







#### **Summary**

- Initial groundwater performance monitoring data looks promising
- U concentrations lowered to below MCL in most wells within a radial distance of 75 ft
- Additional data/evaluation is needed to assess poor performance at two lower zone monitoring wells
- Core samples scheduled for collection in Oct/Nov will provide more definitive performance assessment data